

Recitation Week 6

2/20/13

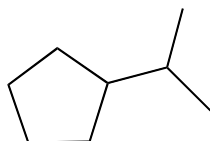
1. Draw the molecules with these names:

(c) 1-octen-7-yne

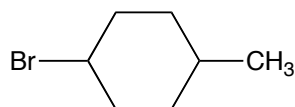
(d) *cis*-3-hexene(e) *trans*-3-hexene

2. Name these molecules

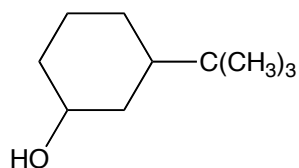
(a)



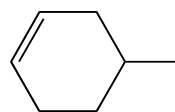
(b)



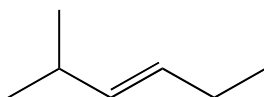
(c)



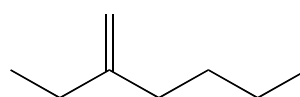
(d)



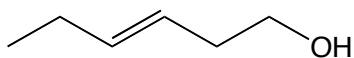
(e)



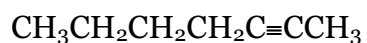
(f)



(g)



(h)



3. How many degrees of unsaturation would a molecule with each of these formulas have? Draw or name an example of a compound with each formula.

(a) C_6H_{12} (b) C_4H_6 (c) $\text{C}_{10}\text{H}_{22}$ (d) $\text{C}_4\text{H}_8\text{O}$ (e) $\text{C}_5\text{H}_8\text{Cl}_2$

4. The following compounds have been named incorrectly. Draw each molecule, and then give its correct IUPAC name.

(a) 4-bromo-3-pentanol

(b) 5-methylcyclohexanol

(c) 2-ethylpentane

(d) 2-methyl-2-penten-4-ol

(e) 5-(2,2-dimethylethyl)decane

5. Draw Newman projections looking down the C_2-C_3 bond showing the most stable conformations of each of the following:

(a) propane

(b) butane

(c) pentane

(d) 2-methylbutane